

Weathering

Weathering - the breakdown of rocks into soil

Types of Weathering:

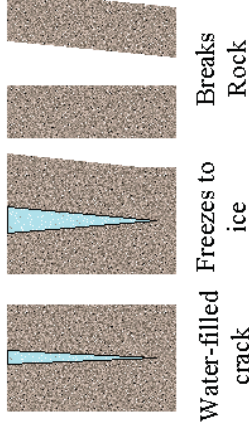
1. **Physical Weathering** - any process that causes a rock to crack or break into pieces without changing it
2. **Chemical Weathering** - any process that causes rocks to breakdown by chemical action
 - results in a change in composition

Physical Weathering

Types of Physical Weathering:

- a. **Frost Action (Ice Wedging)** - water seeps into cracks in a rock

- when water freezes, it expands by 10% causing the rock to split apart



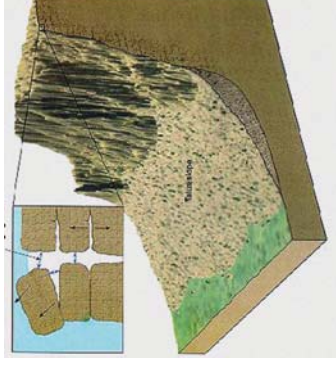
- b. **Extreme Temperature Changes (Exfoliation)** - rocks are heated by the sun and expand; when temperatures fall, the rock cools and contracts

- this cycle of heating and cooling (expansion and contraction) causes the rock to break into slabs



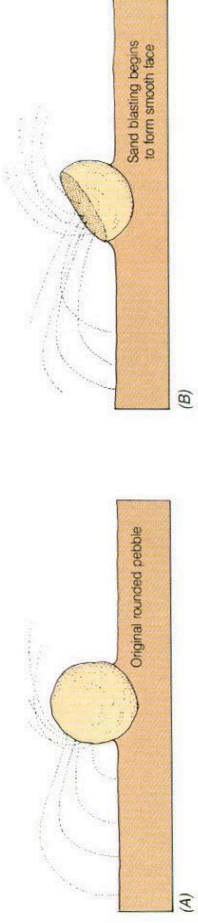
- c. **Plant/Animal Action** - plants/roots will grow into cracks in rocks causing them to split as they grow

- moss and lichens produce acids that weaken rock (chemical breakdown)



Physical Weathering

- d. **Abrasion** - sediments carried by streams and wind blown sand cause particles to collide into each other and the surrounding rock



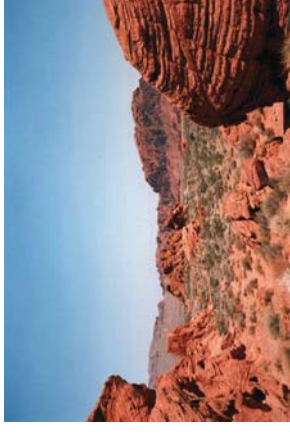
- e. **Pressure Unloading** - as a rock is eroded or glacial ice sheets melt, the rocks below are no longer under pressure
- they release this pressure causing the bedrock to crack



Chemical Weathering

Types of Chemical Weathering:

- a. **Oxidation** - oxygen combines with certain minerals in rocks - the chemical change of the mineral weakens the rock and the rock crumbles
- ex.: rust



- b. **Carbonation** - carbon dioxide dissolves into water and forms a weak acid which reacts with certain rocks and minerals (calcite, limestone, marble, chalk)
- forms sinkholes and caves

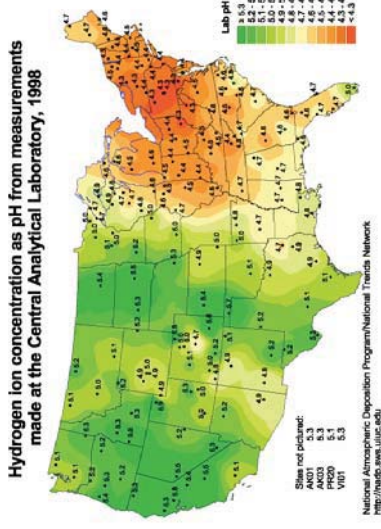
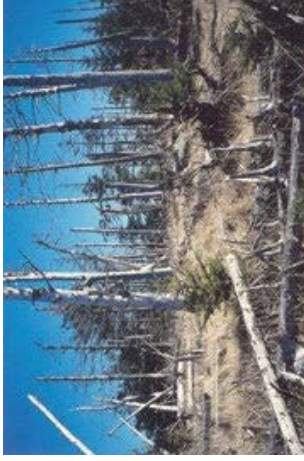


- c. **Hydration** - certain minerals in rocks will dissolve in water and rock will crumble
- ex.: feldspar in granite - feldspar turns to clay



Chemical Weathering

- d. **Acid Rain** - gases released from the burning of fossil fuels dissolve into water droplets in clouds to produce an acid
- ex.: sulfuric acid

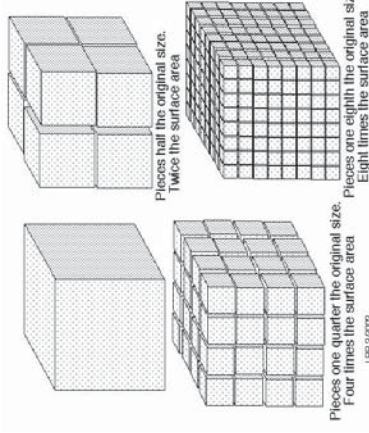


Weathering

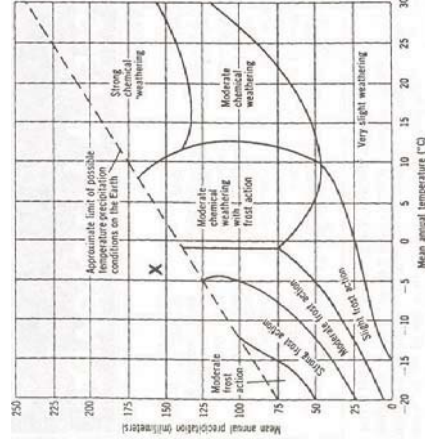
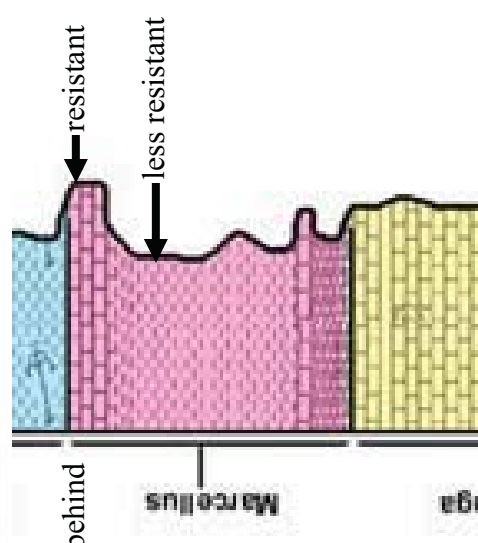
Factors that Effect Weathering:

- Surface Area/Particle Size
- Minerals in Rock
- Climate

- Surface Area** - as surface area increases, weathering increases
 - small particles have more surface area than large particles



- Minerals in Rock** - as hardness of minerals increases, weathering rate decreases
 - softer, less resistant minerals/rocks wear away leaving harder, more resistant minerals/rocks behind



- Climate** - the major factor that effects weathering
 - as humidity increases, weathering increases
 - as temperature increases, chemical weathering increases
 - warm, moist climates (mT)
 - as temperature decreases, physical weathering increases
 - cold, moist climates (mP)

Weathering

Products of Weathering:

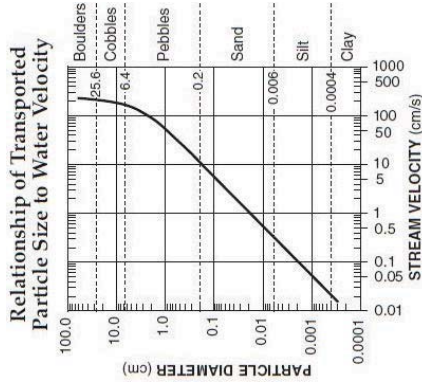
1. Solid Sediments
2. Dissolved Minerals
3. Soils



1. Solid Sediments (from largest to smallest):

- Boulders
- Cobbles
- Pebbles
- Sand
- Silt
- Clay
- Colloids

Reference Tables p.6

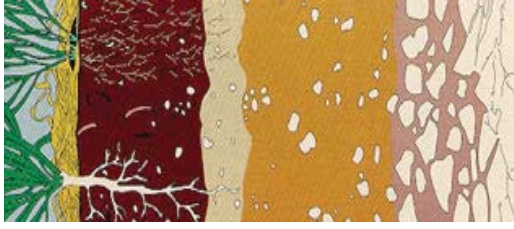


- colloids are the smallest particles and always remain suspended in water - never settle out

2. Dissolved Minerals - cause "hard" water

- when water evaporates, dissolved minerals will precipitate out and settle to the bottom

3. Soil - combination of weathered rock and organic matter (**humus** - decayed plant/animal remains)



Topsoil - contains humus

Subsoil - contains leached minerals

C-Horizon - partially weathered bedrock

Bedrock - often the parent rock of soil above

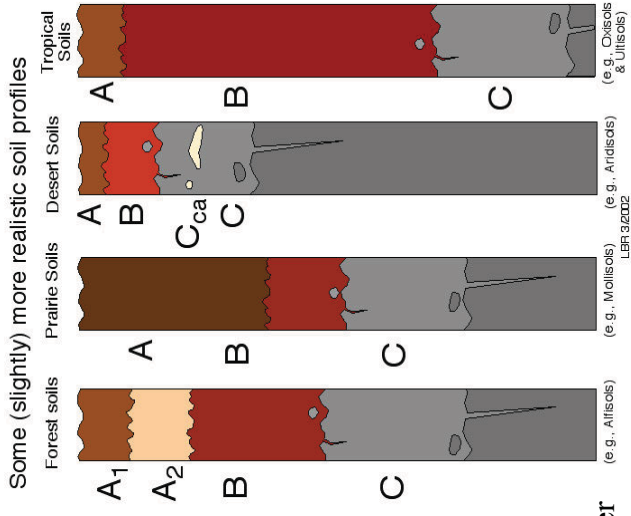
Sediments



Types of Soils:

1. **Residual Soils** - weathered rocks/particles are the same as the underlying bedrock
2. **Transported Soils** - weathered rock/particles do not match the underlying bedrock (transported from elsewhere)
 - ex.: soils in NYS formed from rocks that came from Canada and were transported by glaciers and deposited in NYS during the last ice age

- soil profiles that form in different environments will have very distinct differences from each other



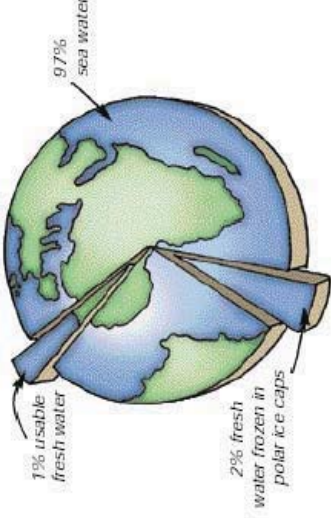
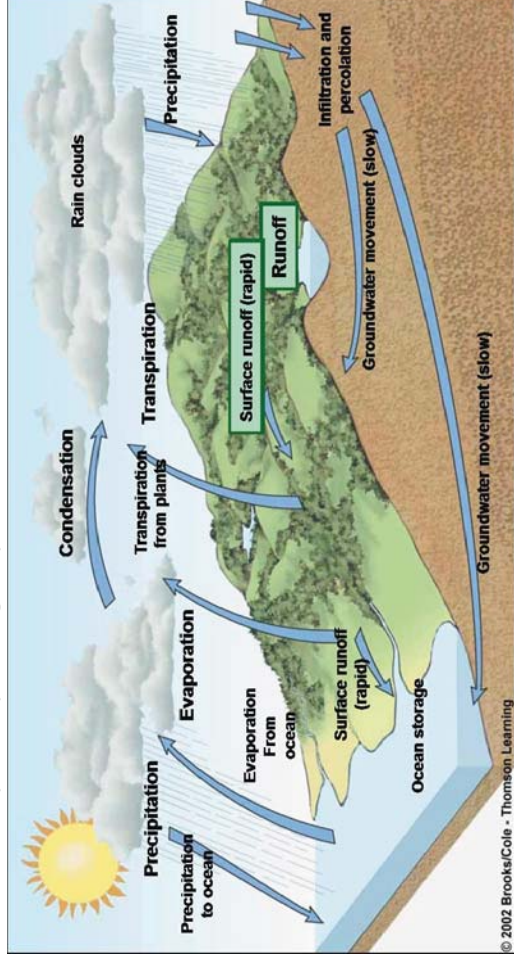
Water Cycle

Most of Earth's water was created by volcanic out-gassing when the Earth formed and by impacts by comets early in Earth's history

- this water has remained since – the amount of water on Earth has not changed
- the majority of fresh water is unusable – frozen in glacial ice

Water on Earth is recycled from the atmosphere to the Earth and back to the atmosphere

- a model used to show the movement and phase changes of water at or near the Earth's surface is the **water cycle (hydrologic cycle)**:



The Sun is the source of energy for the water cycle

- the source of most water on land is the oceans
- water from the atmosphere reaches land through the process of **precipitation**

When precipitation reaches land it will:

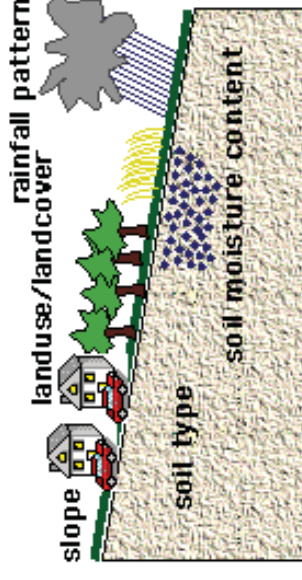
1. **infiltrate** into the ground – stored as groundwater
2. **runoff** the surface into streams, lakes and the ocean
3. stored at the Earth's surface as ice/snow
4. evaporated and transpired (**evapotranspiration**) back into the atmosphere

Infiltration

Infiltration – process of water sinking into the ground under the influence of gravity

Factors Affecting Infiltration:

1. slope of the land
2. porosity
3. permeability
4. degree of saturation
5. capillarity
6. vegetation/land use

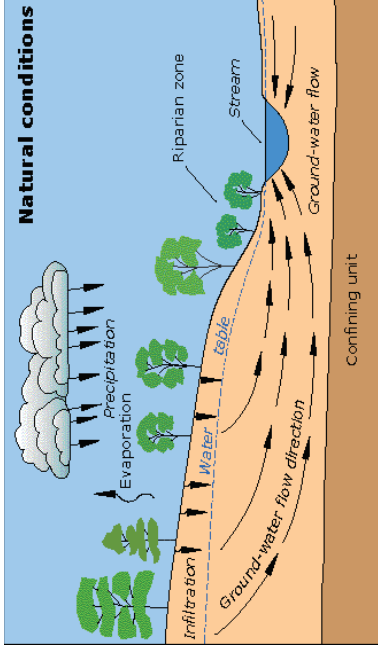


Major factors affecting infiltration

Slope of the Land

The steeper the slope of the land, the less water will infiltrate into the ground – water will runoff the surface

- the gentler the slope, the more water will infiltrate

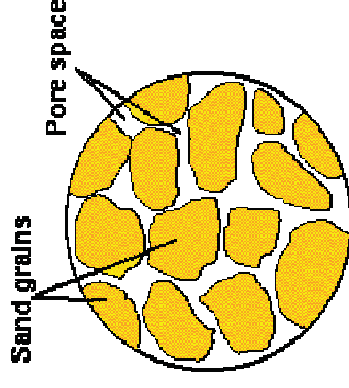


Porosity

Porosity – the percentage of open air space in a material

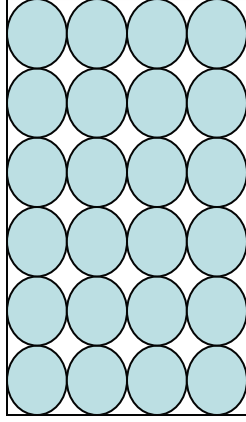
Factors that Effect Porosity:

1. shape of particles
2. packing of particles
3. sorting of particles

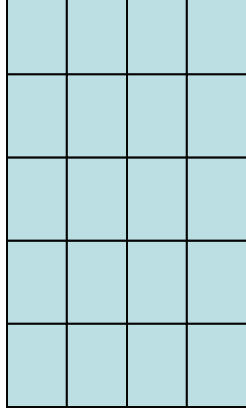


Shape of Particles – well rounded particles have greater porosity than angular or flat particles

- round particles do not fit together as tightly as other shapes



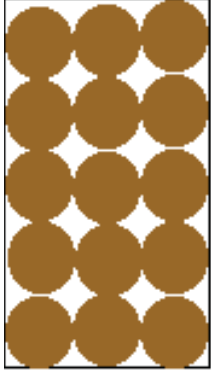
lots of pore space



little/no pore space

Porosity

Packing of Particles – the more closely packed the particles are, the lower the porosity



loosely packed,
uniform particle size

lots of pore space

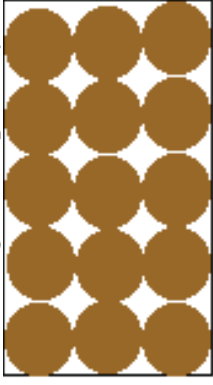


tightly packed,
uniform particle size

little pore space

Sorting of Particles – if all the particles of a material are of the same size, they are **sorted**; if they are different sizes, they are **unsorted**

- the more sorted the particles, the higher the porosity



loosely packed,
uniform particle size

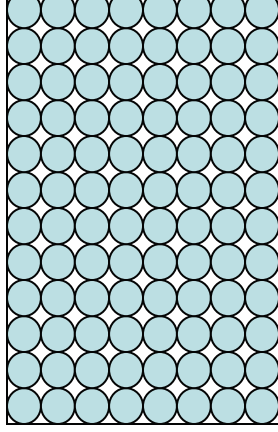
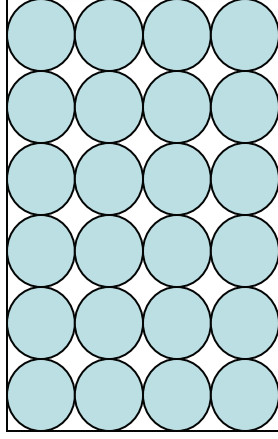
lots of pore space



tightly packed,
mixed particle sizes

little pore space

The size of particles **DOES NOT** effect porosity!!



- each of the samples have round particles that are sorted and are packed similarly
- each sample has the same porosity: ~40%

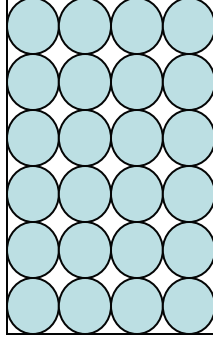
Permeability

Permeability – how easily a material allows water to pass through it

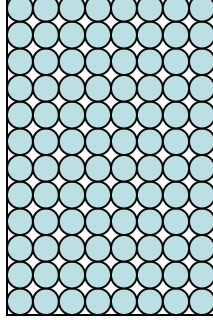
Factors that Effect Permeability:

1. size of particles
2. shape of particles
3. packing of particles
4. sorting of particles

Size of Particles – the bigger the particle size, the easier water can flow through a material



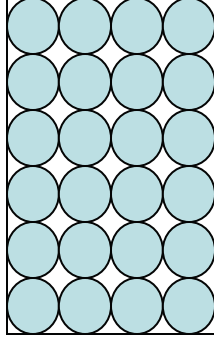
high permeability



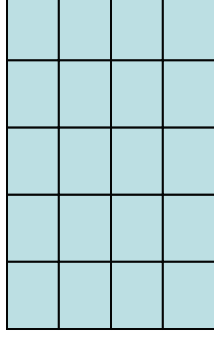
low permeability

- water has to move around/through many more particles for the same volume of material – takes longer

Shape of Particles – the rounder the particles are, the more pore space a material has and the easier water will flow through a material

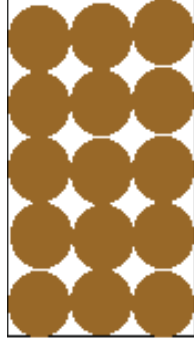


high permeability



low permeability

Packing of the Particles – the more tightly packed the particles of a material are, the harder it is for water to flow through the material



loosely packed,
uniform particle size

high permeability



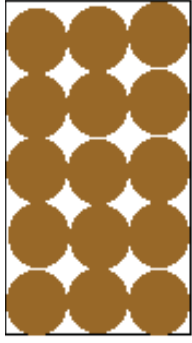
tightly packed,
uniform particle size

low permeability

- loosely packed particles have more pore space

Permeability

Sorting of Particles – the better sorted the particles of a sample are, the easier it is for water to flow through the material



loosely packed,
uniform particle size

high permeability



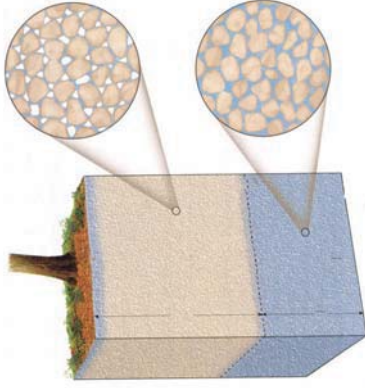
tightly packed,
mixed particle sizes

low permeability

Degree of Saturation

A material is **saturated** when the pore spaces in the material are completely filled with water

- the more saturated a material is, the less the infiltration of water there will be

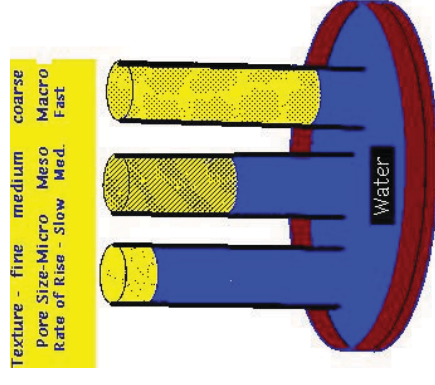
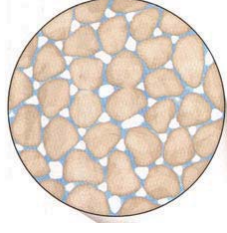


- has open pore spaces – water can flow easily
- pore spaces are filled with water – more water will not pass through

Capillarity

Capillarity – the attractive force of water molecules and particles of sediment

- causes some water to move upward against gravity within a material
 - smaller particles have a greater capillarity than large particles
 - the finer the particles of a material are, the faster and farther water can move upward through the material



Vegetation/Land Use

Grasses, plants and trees will slow water flow on the surface of the Earth – allows water time to infiltrate into the ground

- exposed ground/bedrock usually results in water easily running off the surface
- buildings, roads, parking lots etc. are impermeable - reduce infiltration
- farming and deforestation also reduce infiltration



Runoff

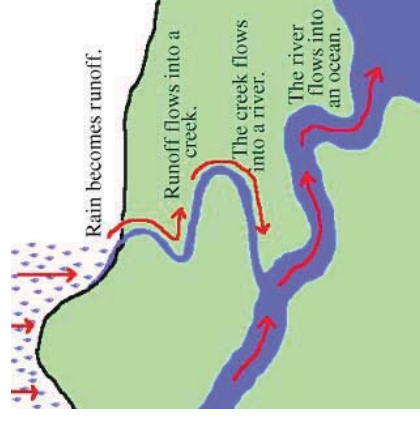
Runoff – water that flows on the Earth's surface

Runoff occurs when:

1. precipitation is greater than infiltration
 2. ground is saturated
 3. slope of the surface is great
- runoff eventually reaches streams

Stream Discharge – the volume of water flowing in a stream during a given amount of time

- the greater the amount of runoff, the greater the stream discharge



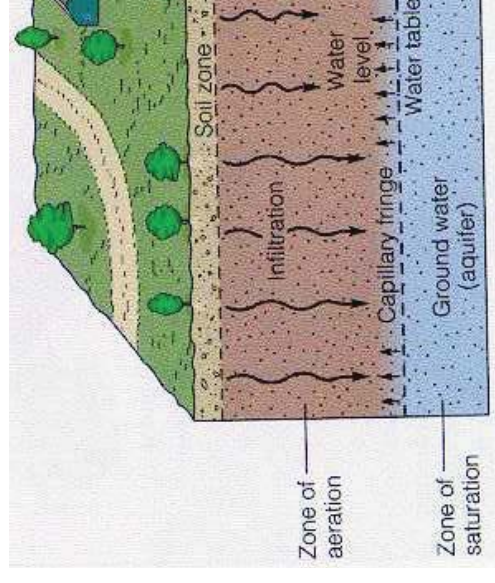
Groundwater

Earth's surface is divided into two zones:

Zone of Aeration – pore spaces are filled with air or are partially filled with water due to capillary action

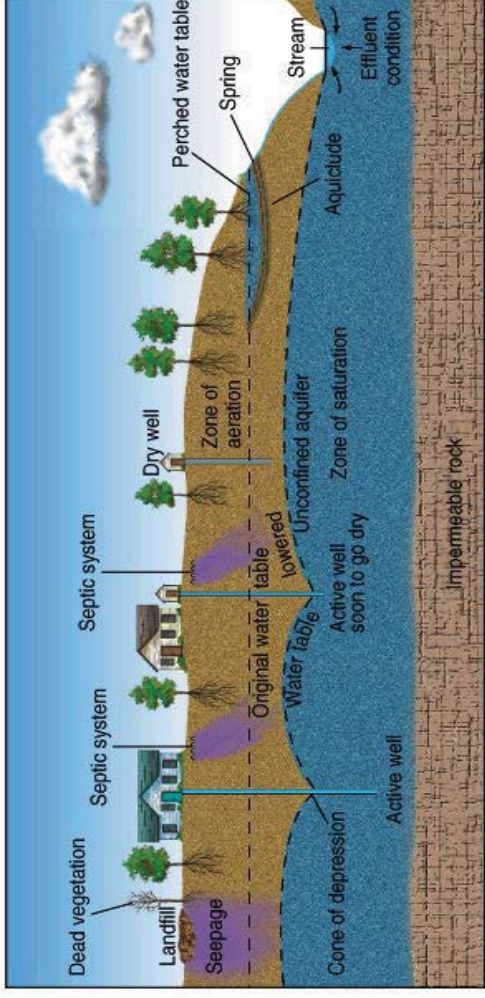
Zone of Saturation – pore spaces are filled with water

- the water held in the ground within this zone is called **groundwater** (water will flow to streams, rivers, etc.)
- The interface between these two zones is called the **water table**
- **aquifer** – a source of groundwater used as a drinking supply

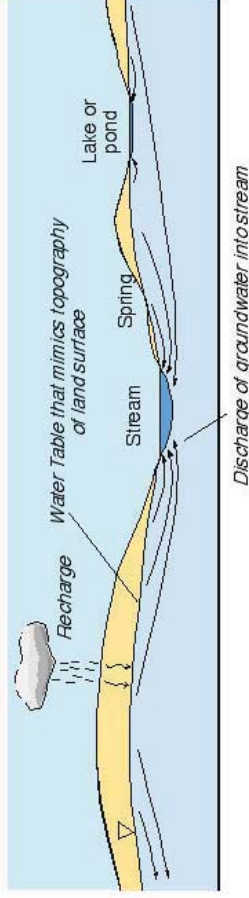


Groundwater

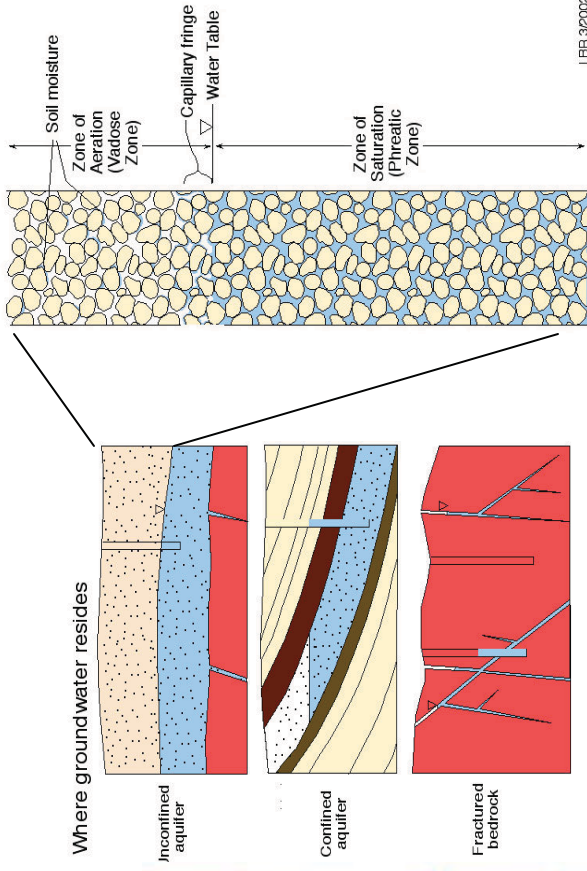
- any substantial source of groundwater can be used as an aquifer
- the water table will lower as wells withdraw water faster than can be replaced by infiltration



- the water table will roughly mimic the contours of the land surface:



- streams, rivers, lakes, ponds, and marshes are intersections of the water table with the land surface
- groundwater moves down slope due to gravity
- groundwater discharges into streams and the ocean – most water in streams and rivers comes from groundwater
- extensive pumping of the groundwater will lower the water table
 - this can lead to a “dry well”



LBR 3/2002

